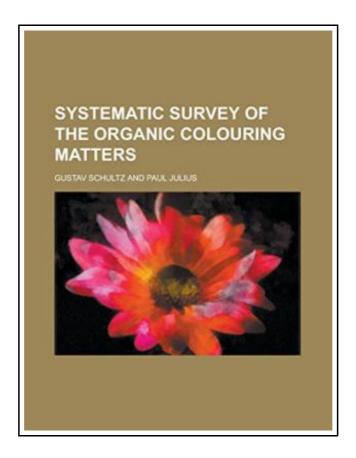
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SYSTEMATIC SURVEY OF THE ORGANIC COLOURING MATTERS (PAPERBACK)



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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English. Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1894 edition. Excerpt: .acids which are formed by further sulphonation of anaphtholmonosulphouic acids (1:4) and (1:5) (Bayer and Co. Eng. Pat. 339790). Dioxynaphthalene (2:0) noh Dioxynaphthalene (2:7) HO AMIDOPHENOLS AND THEIR DERIVATIVES o-Anisidine (Methyl Ether of o-Arnidophenol) OCH, ONH2. Preparation.--By reduction of the methyl ether of w-nitrophenol (Miilhhiiuser, Ann. 207, 239). Properties.--(r. B.p. 226 at 734 mm. Preparation.--(1) By fusing m-sulphanilic acid (1 pt.) with caustic soda (2 pts.) and a little water at 280 to 290. The melt is dissolved in water, acidified with hydrochloric acid, sodium carbonate added, and the solution extracted with ether or aniyl alcohol. The crude product obtained after distilling off the solvent is purified by crystallisation from hot water (Gesellsch. f. Chem. Industrie, Ger. Pat. 44792; Am. Pat. 403678). (2) By heating resorcinol (10 pts.) with aqueous ammonia of 10 NH3 (20 pts.) and ammonium chloride (6 pts.) for 12 hours at about 200 in an autoclave (Leonhardt and Co., Eng. Pat. 815689). Properties.--White prismatic needles. M.p. 121. Easily soluble in hot water, moderately in cold. Dimethyl-mamidophenol (m-Oxydiuidhylaniline) C6H4(OH).N(CH:,),1:3 Preparation.--(1) By soda fusion (v. supra) of dimethyl-w-sulphanilic acid, obtained by sulphonating dimethylaniline with fuming sulphuric acid (Gesellsch. f. Chem. Industrie, Ger. Pat. 44792s8). (2) By heating resorciuol with aqueous dimethylamine and dimethylamine hydrochloride at about 200 in an autoclave (Leonhardt and Co., Eng. Pat. 815689). Properties.--Crystals. M.p. 87; b.p. 265 to 268. Nearly insoluble in water. By nitrous acid it is converted into a nitroso compound C(.H3(NO) (OH)-N(CH3)., the hydrochloride of...



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